



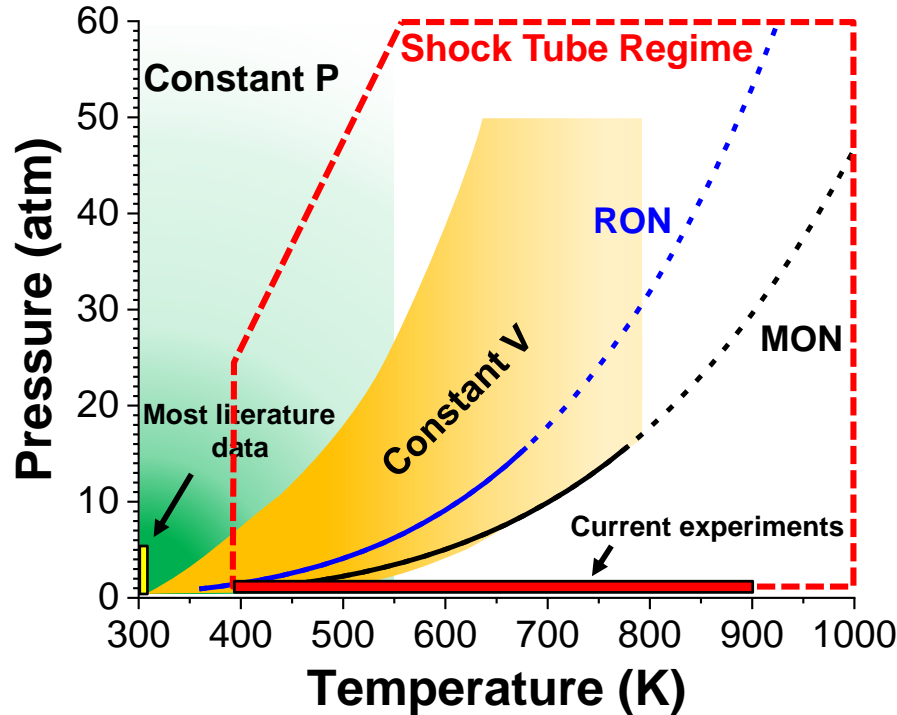
High-temperature laminar burning velocity experiments in a shock tube: LBV, temperature, and species measurements

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Motivation: Why study flames in a shock tube?



- Limited flame speed data at high unburned gas T (>500 K)
- Shock tubes are ideal reactors for studying high-temperature combustion phenomena
 - › Tunable T & P
 - › Near-instantaneous heating (well-defined time zero)

Overview of Experiments

Goal: Use a shock tube to extend the temperature regime accessible by existing laminar flame speed measurement approaches

Laminar flame speed experiments in a shock tube

I. Small Hydrocarbons

High-temperature flame speed measurements:
 CH_4 , C_2H_6 , C_3H_8

II. Laser Diagnostics

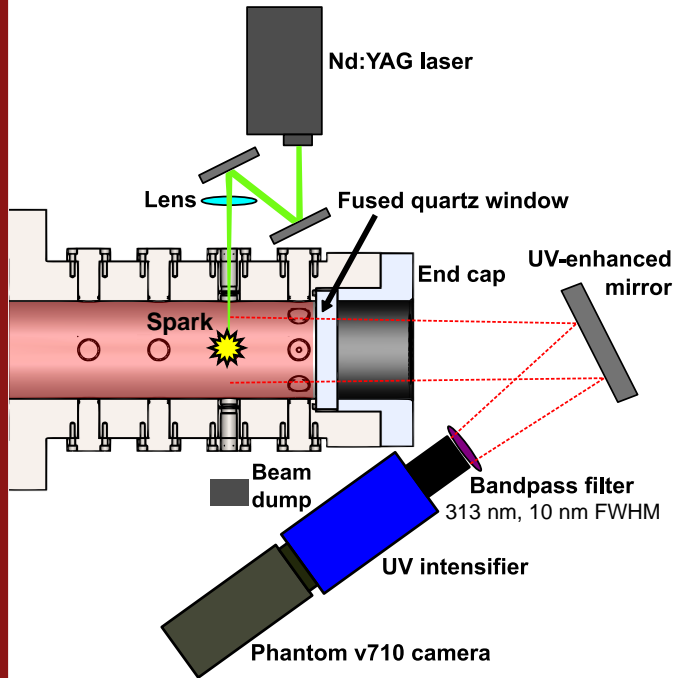
Temperature and species (CO_2 & H_2O) measurements in C_2H_6 -air flames

III. NTC Regime

iso-octane flame speed measurements in the NTC regime

First, the approach...

Experimental Setup: Laminar flames in a shock tube



Laminar flames are initiated behind reflected shock waves via laser-induced spark ignition

Shock tube

- 11.53-cm inner diameter
- 3.63-m driver, 9.73-m driven
- Test times: 3.5-15 ms (N_2 , He, CO_2 driver)

Nd:YAG laser

- 532 nm, 5 ns pulse, ~18 mJ/pulse

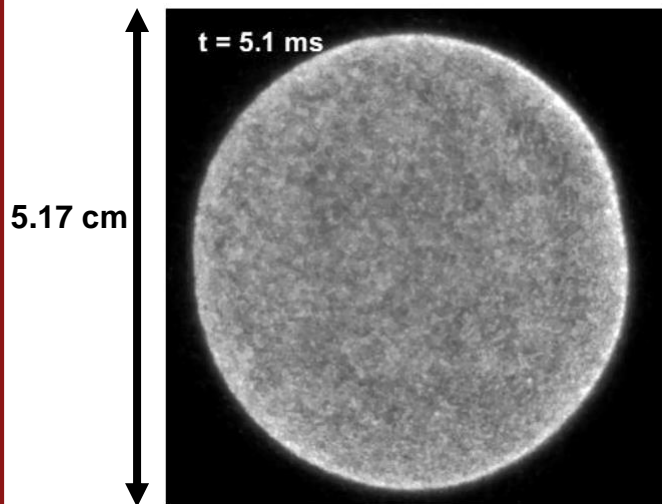
High-speed OH* emission (306 nm) imaging

- UV-intensified, endwall emission

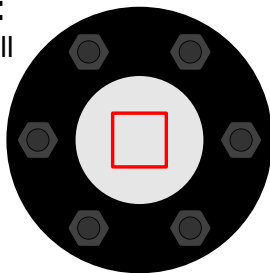
OH* Emission Images

Example: raw images (OH* emission, 306 nm)

C₂H₆ in air, $\Phi = 1.0$, 488 K, 1.0 atm



Field of View:
Shock tube endwall



- UV-intensified OH* emission imaging
- 10-40 kHz frame rate
- 90-140 $\mu\text{m}/\text{pixel}$
- Field of view:
 - › 5.17 x 5.17 cm (384 x 384 pixels)
 - › 10 cm from shock tube endwall

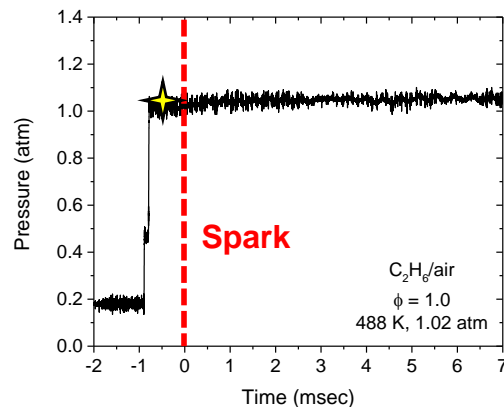
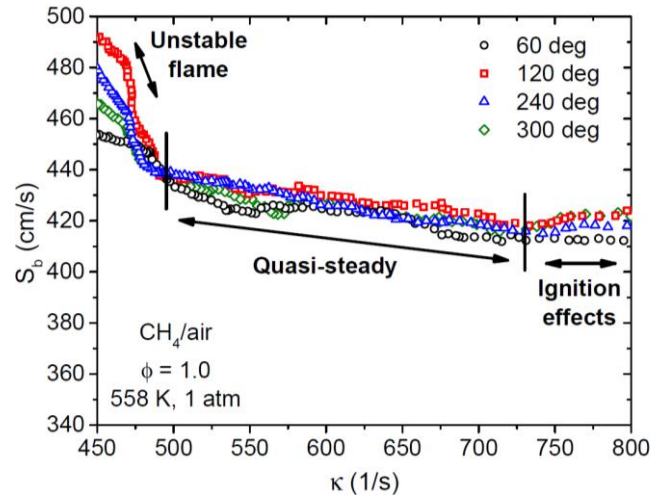


Image Processing



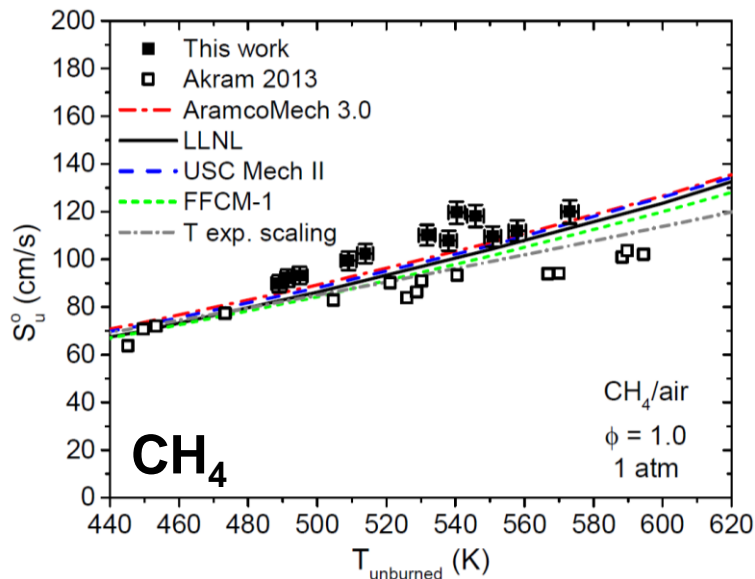
- Local stretch rate (κ) and burned flame speed (S_b) calculated at regular intervals around circumference of flame
- Quasi-steady subset of S_b vs κ data selected for unstretched flame speed extrapolation [1]

$$S_b^2 - S_b^0 S_b = -S_b^0 L_b \kappa \quad \rho_u S_u^0 = \rho_b S_b^0$$

I. Small Hydrocarbons

CH₄ and C₂H₆ flame speed measurement results:

Mixture	CH ₄ /air	C ₂ H ₆ /air
Φ	1.0	1.0
T _{unburned} (K)	489-573 K	449-537 K
Pressure (atm)	1.0 ± 0.05	1.0 ± 0.05



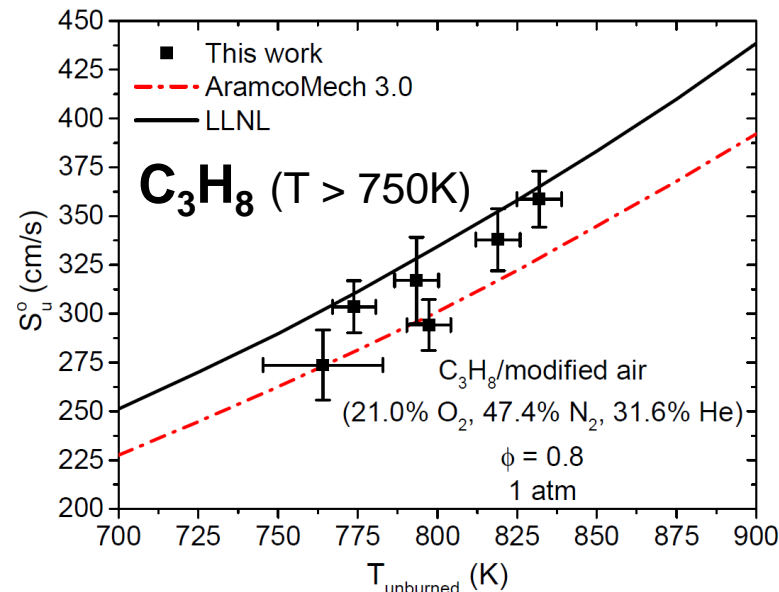
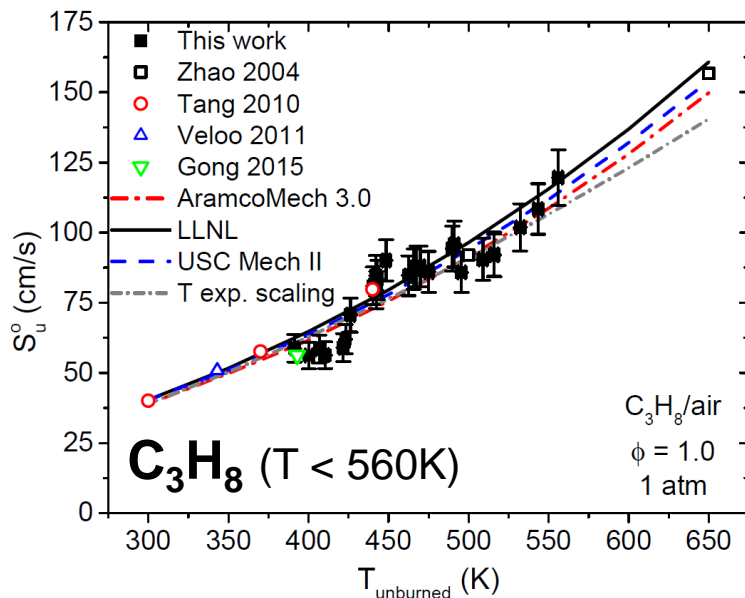
The high-temperature C₂H₆ flame speed results have not yet been published, and have therefore been removed from the publicly distributed version of this talk.

- CH₄ results are 5-10% higher than model results
 - Only one other high-T CH₄ data set in literature
- C₂H₆ results show close agreement with USC Mech results, higher disagreement with Aramco results

I. Small Hydrocarbons

Mixture	C ₃ H ₈ /air	C ₃ H ₈ /modified-air
Φ	1.0	0.8
T _{unburned} (K)	391-556 K	764-832 K
Pressure (atm)	1.0 ± 0.05	1.0 ± 0.05

C₃H₈ flame speed measurement results:

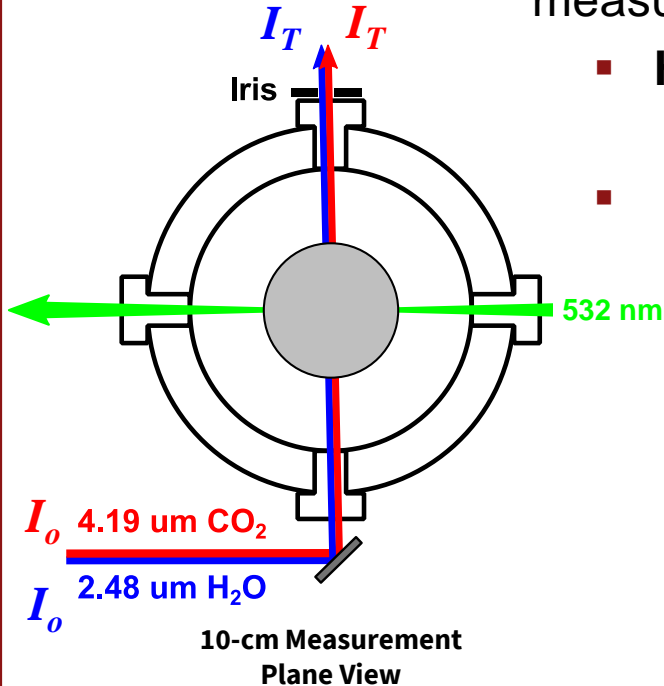


- Results show excellent agreement with model and (available) literature results

- Disagreement between model results at higher temperatures highlights need for high-T flame speed validation data

II. Laser Diagnostics

- Two time-multiplexed, scanned-direct absorption diagnostics were used to measure temperature, CO₂, and H₂O in C₂H₆-air flames:



- **H₂O:**

- › 1 transition near 2.48 μm (4029.52 cm⁻¹) [2]

- **CO₂ and Temperature:**

- › 2 transitions near 4.19 μm (2384.19, 2384.33 cm⁻¹) [3]

Beer-Lambert Relation:

$$\alpha_v = -\ln\left(\frac{I_T}{I_o}\right) = \frac{X_{Abs} P}{RT} \sigma_v L$$

Labels for the equation:

- α_v : Absorbance
- I_T : Transmitted
- I_o : Incident
- X_{Abs} : Mole fraction
- P : Path length
- σ_v : Absorption cross-section
- L : Path length

- › OH* emission images used to determine time-dependent path length, L

II. Laser Diagnostics: T measurement results

Note: the temperature laser absorption results have not yet been published; the corresponding plots have therefore been removed from the publicly distributed version of this talk.

Time-resolved temperature measurements show:

- Burned gas temperature increases for the first 5-6 ms, then ultimately plateaus to a final, equilibrium burned gas temperature
- Constant long-time temperature indicates minimal radiative losses

Equilibrium burned gas temperature measurements, measured for each ethane-air flame experiment (1 atm, 449-537 K), show:

- Excellent agreement with AramcoMech 3.0 and USC Mech II modeled results across the entire temperature range (within $\pm 3.2\%$ uncertainty)

II. Laser Diagnostics: Mole fraction results (equilibrium)

Note: the CO₂ and H₂O laser absorption results have not yet been published; the corresponding plots have therefore been removed from the publicly distributed version of this talk.

Equilibrium CO₂ mole fraction measurements show:

- Good agreement with model results (AramcoMech 3.0, USC Mech II), within uncertainty bounds ($\pm 9.2\%$)

Equilibrium H₂O mole fraction measurements show:

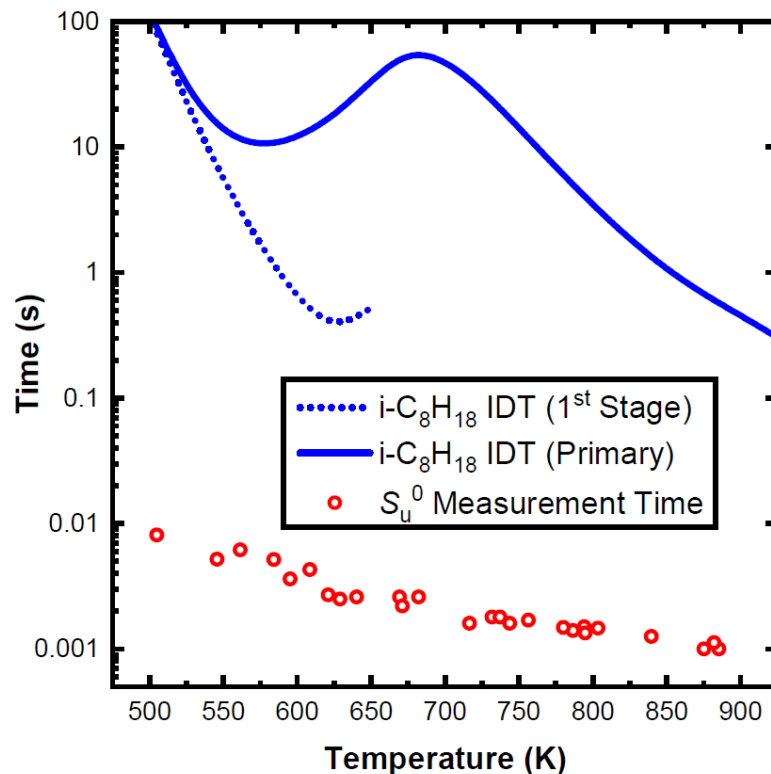
- Satisfactory agreement with model results, within uncertainty bounds ($\pm 10\%$)
 - › Higher scatter likely due to background signal fluctuations

III. NTC Regime: *iso*-octane

iso-octane burning velocity measurements in O₂/N₂/He:

- $\Phi = 0.9$ fuel in 18% O₂, 41% N₂, 41% He
- 500-900 K, 1 atm ($\pm 5\%$)

**IDT \gg experiment time:
no pre-flame reactions
expected in bulk gas**



III. NTC Regime: *iso*-octane

Iso-octane burning velocity measurements in O₂/N₂/He:

- $\Phi = 0.9$ fuel in 18% O₂, 41% N₂, 41% He
 - 500-900 K, 1 atm ($\pm 5\%$)
1. Burning velocity increases w/ T (500-600 K)
 - › Results 20-30% low of simulation
 2. Burning velocity T dependence stronger than modeled (600-750 K)
 3. Negative T dependence of burning velocity (750-820 K)
 4. Excellent model agreement, then stronger T dependence than modeled (820-900 K)

The NTC-regime *iso*-octane flame speed results have not yet been published, and have therefore been removed from the publicly distributed version of this talk.

The NTC-regime *iso*-octane flame speed results show different behaviors over 4 distinct temperature ranges

First direct experimental evidence of NTC flame speed behavior

Summary & Future Work

Summary:

- High-temperature CH_4 , C_2H_6 , and C_3H_8 flame speeds show close agreement with literature and model results
- Laser absorption diagnostics were successfully deployed to measure temperature, CO_2 , and H_2O
- *iso*-octane flame speed results show evidence of NTC behavior – the first such experimental observation

Future work:

- Extend high-temperature flame speed data sets
 - › Additional temperatures, pressures, equivalence ratios
- Use quantitative laser diagnostics to probe NTC flame speed behavior (cool flames?)
- Simulate results as spherically expanding flames using DNS

Acknowledgements

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Questions?