

# New targets for laminar flame speed determination and kinetic schemes assessment

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# 1. INTRODUCTION

# Introduction

## 2. SET-UP

2.1 Flame visualisation

2.2 Rf and P trace

## 3. Validation & limits

3.1 Heat losses

3.2 Stretch

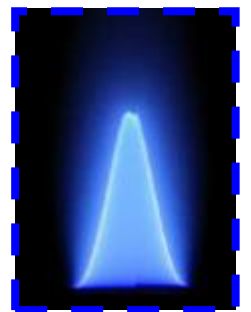
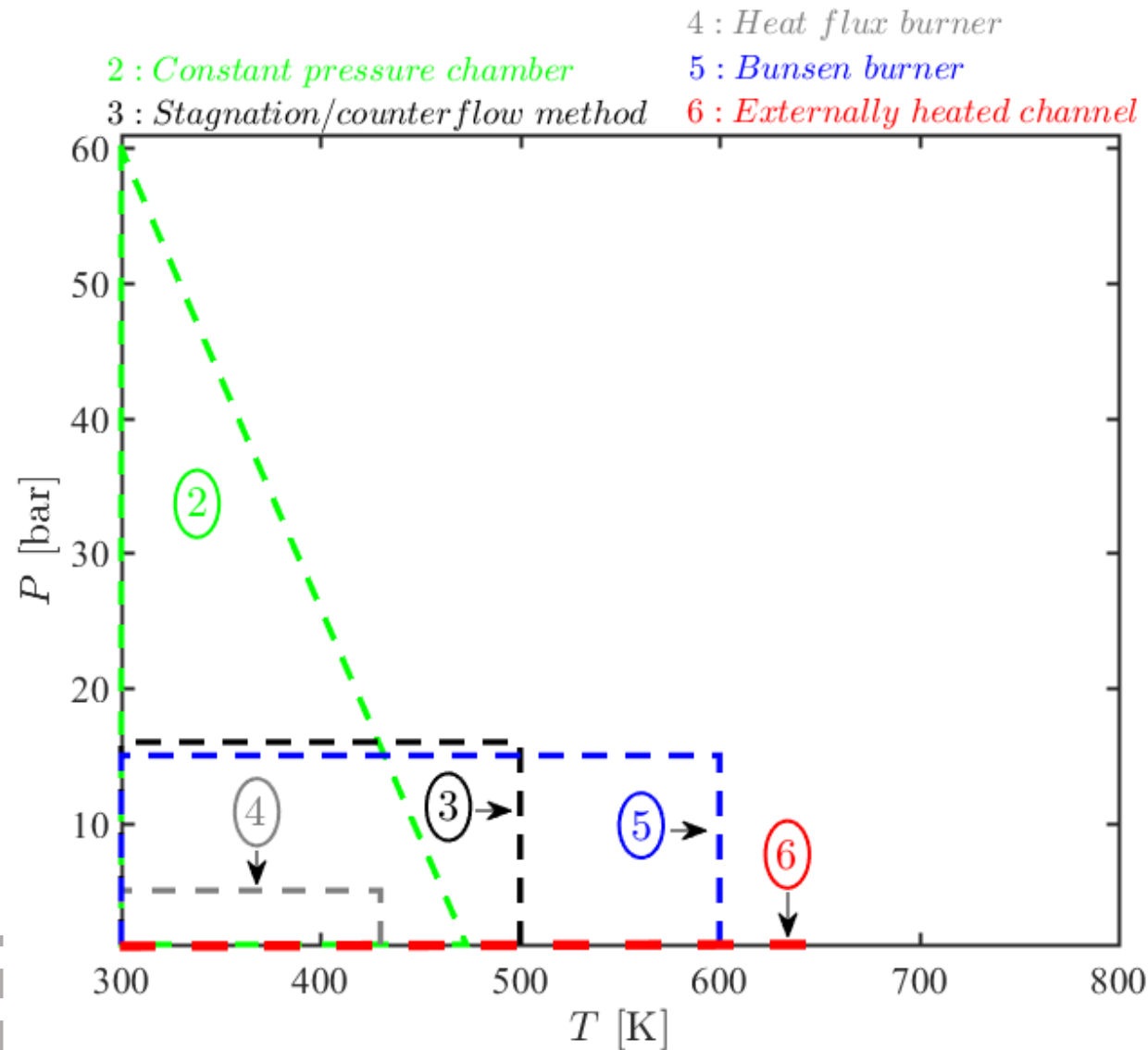
## 4. RESULTS

4.1 Flame speed

4.2 New target

4.3 New method

## 5. CONCLUSIONS



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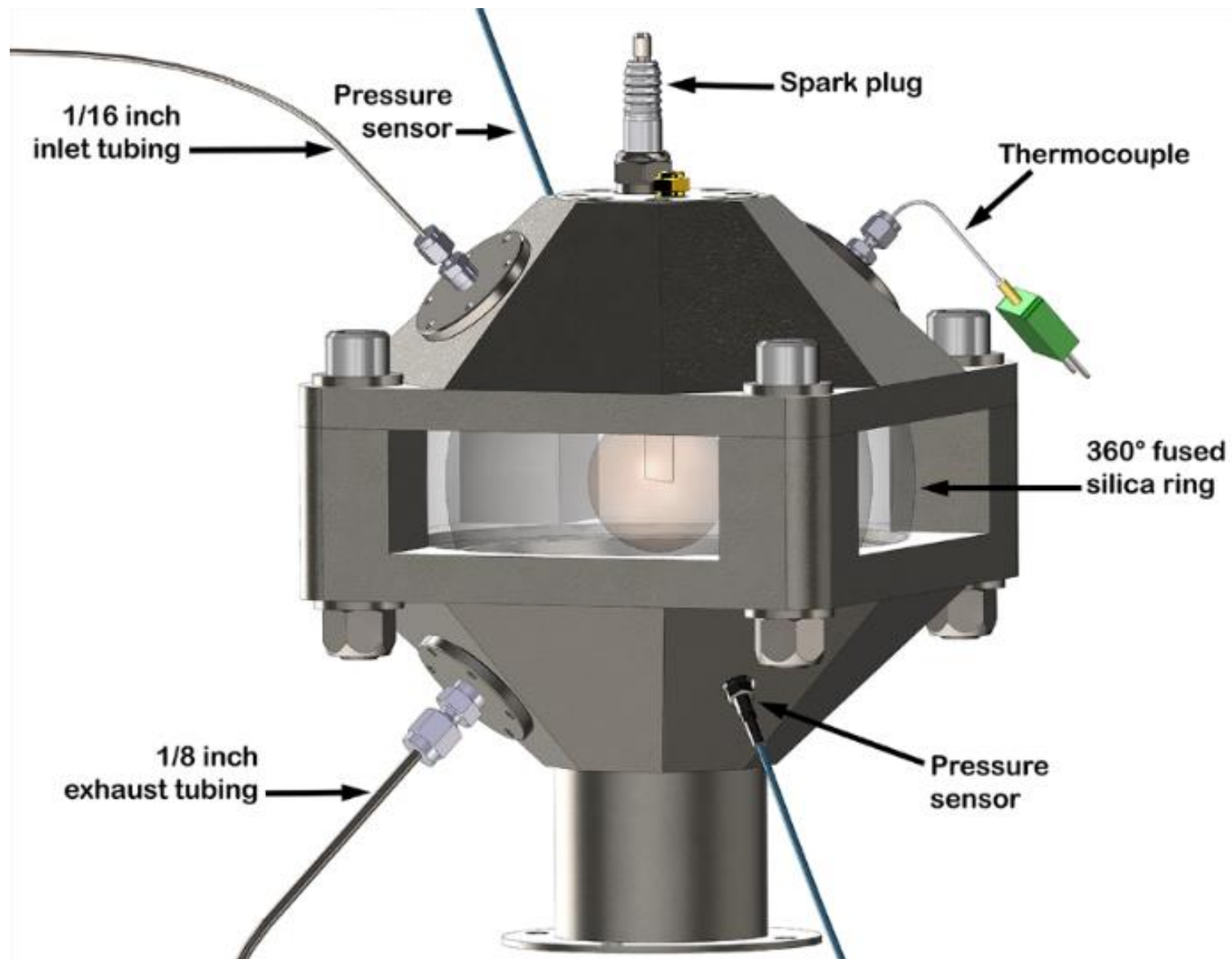
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# full OPTical access Perfectly spheriCal combustIon chaMber (OPTIPRIM)



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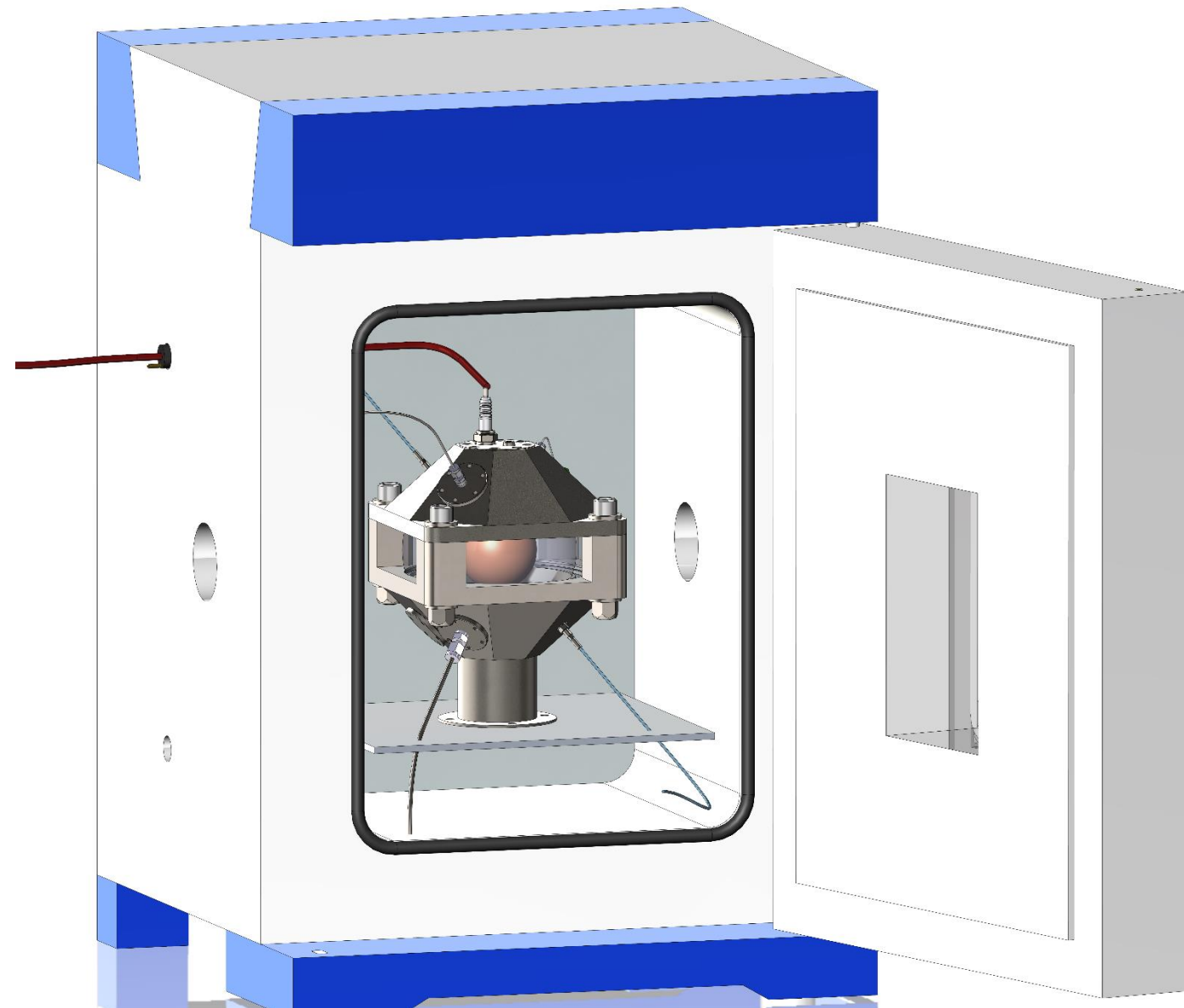
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# Experimental set-up

Fuel	$T_0$ (K)	$P_0$ (bar)	$\phi$ (-)
$CH_4$	300	1	1
$CH_4$	300	1	1.3



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# Flame propagation

CH<sub>4</sub>/air at  $\phi = 1.0$



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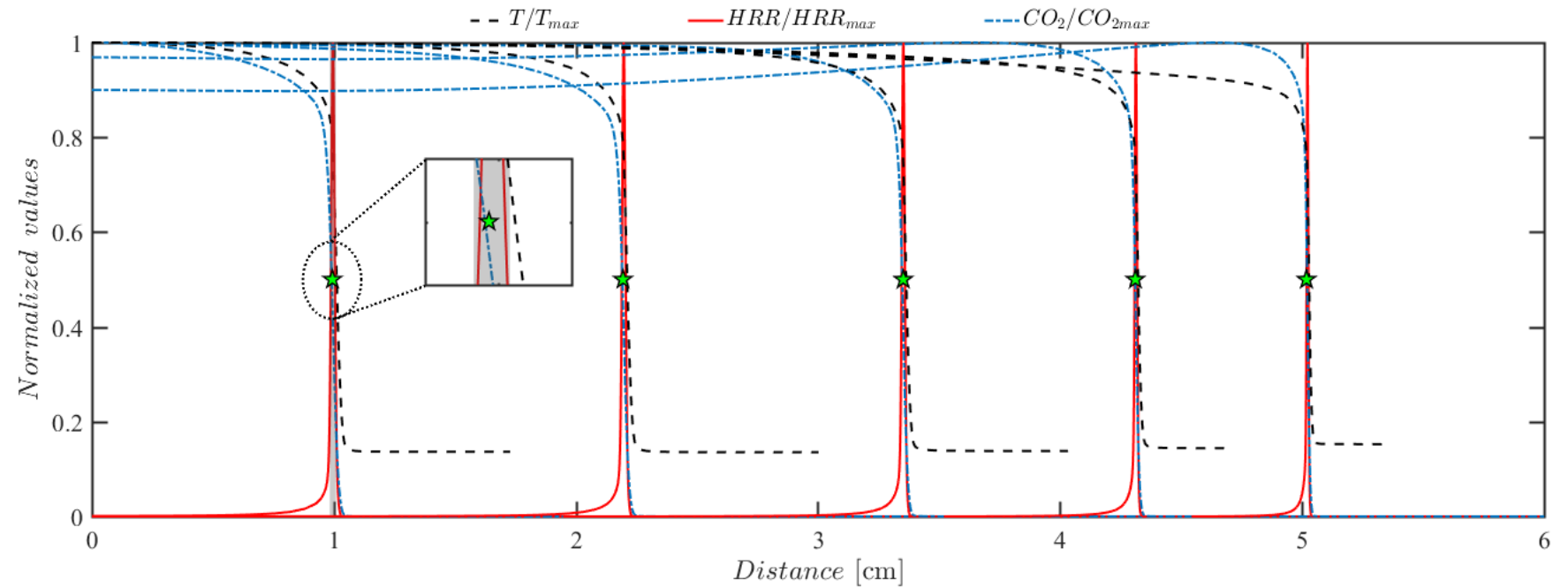
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6 ms

11 ms

16 ms



21 ms

26 ms

27 ms



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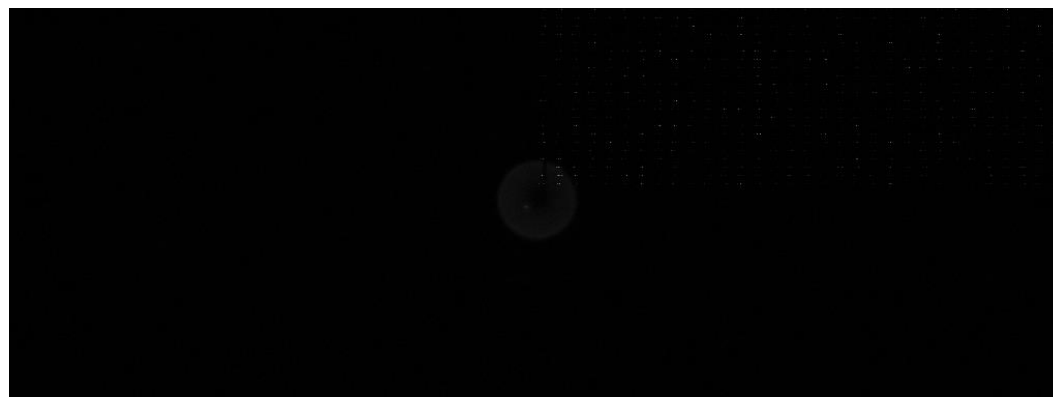
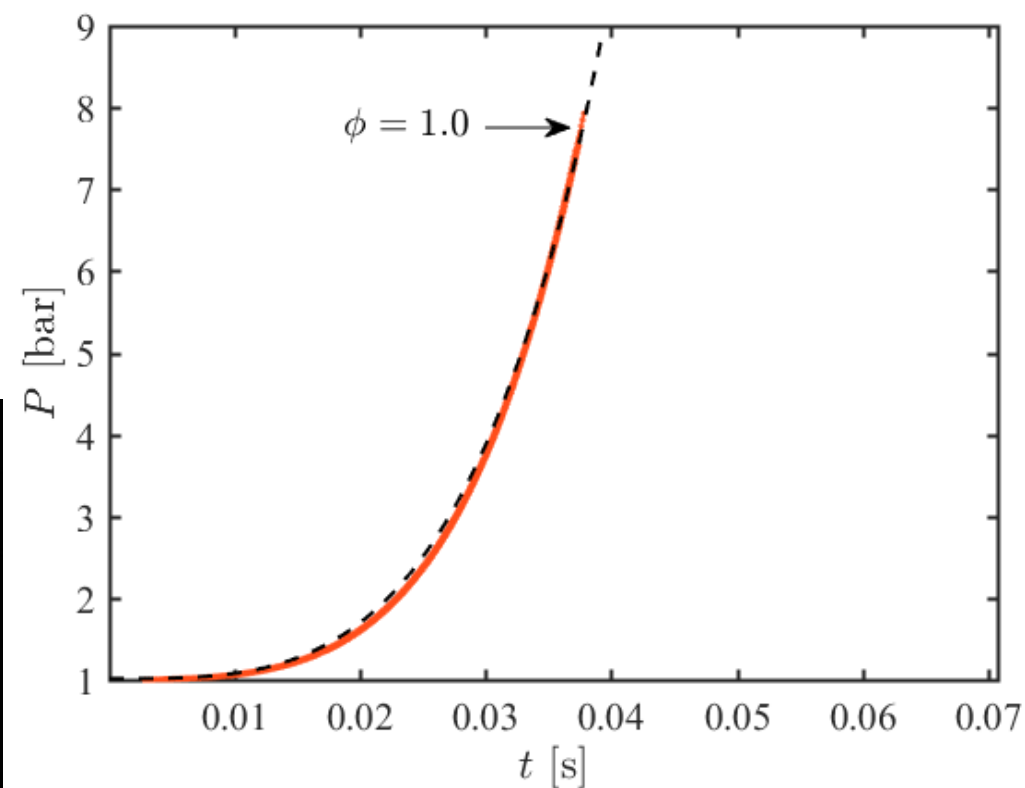
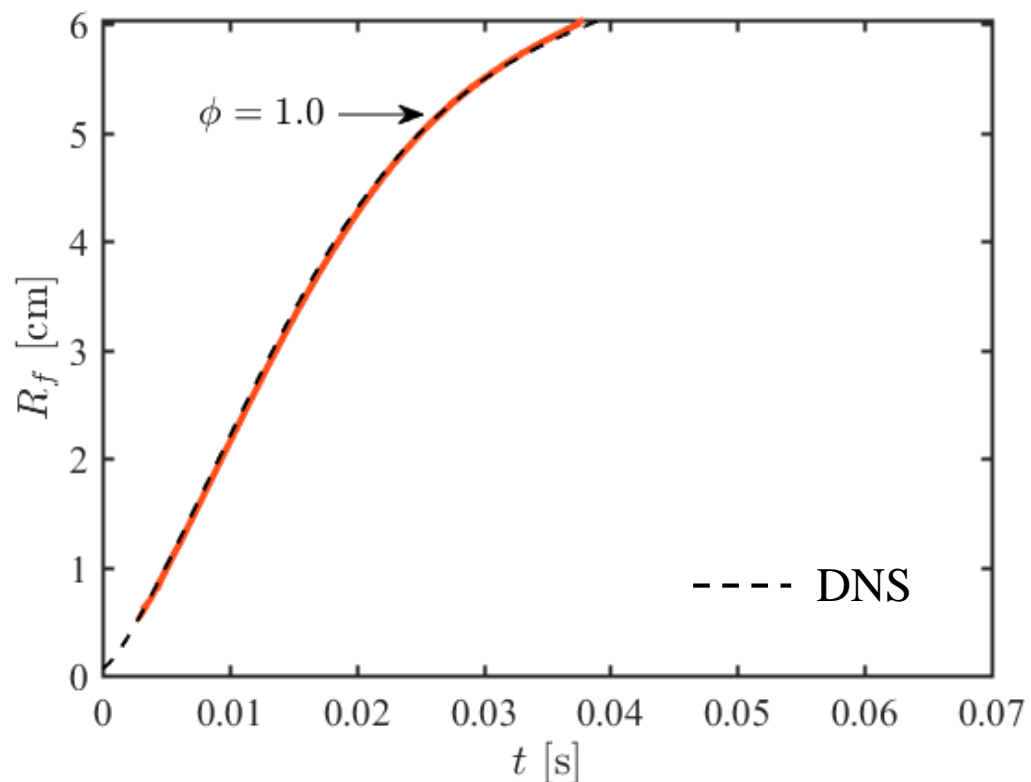
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# Flame radius & pressure evolutions



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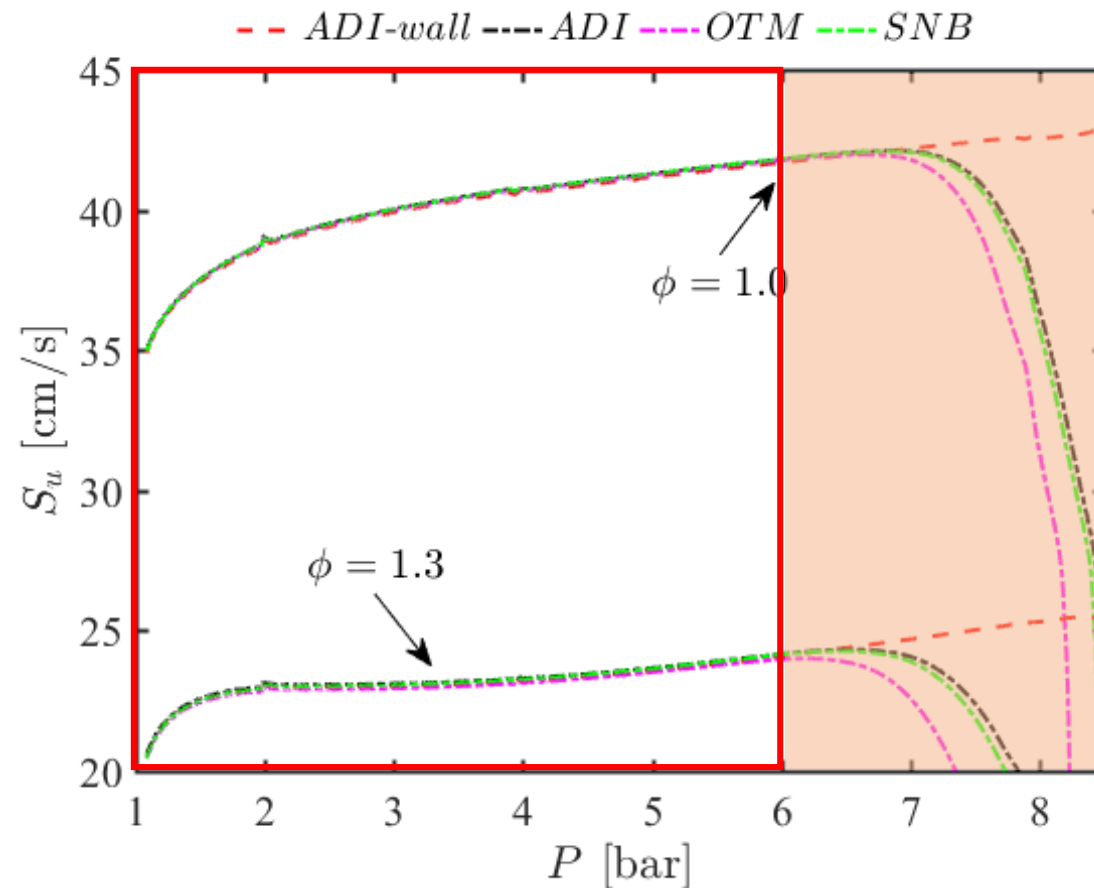
# Radiation and heat losses to the walls

**ADI-wall** adiabatic walls

**ADI** adiabatic model with no radiative loss

**OTM** optically thin model considering emission but no absorption

**SNB** statistical narrow band model with both radiation emission and absorption





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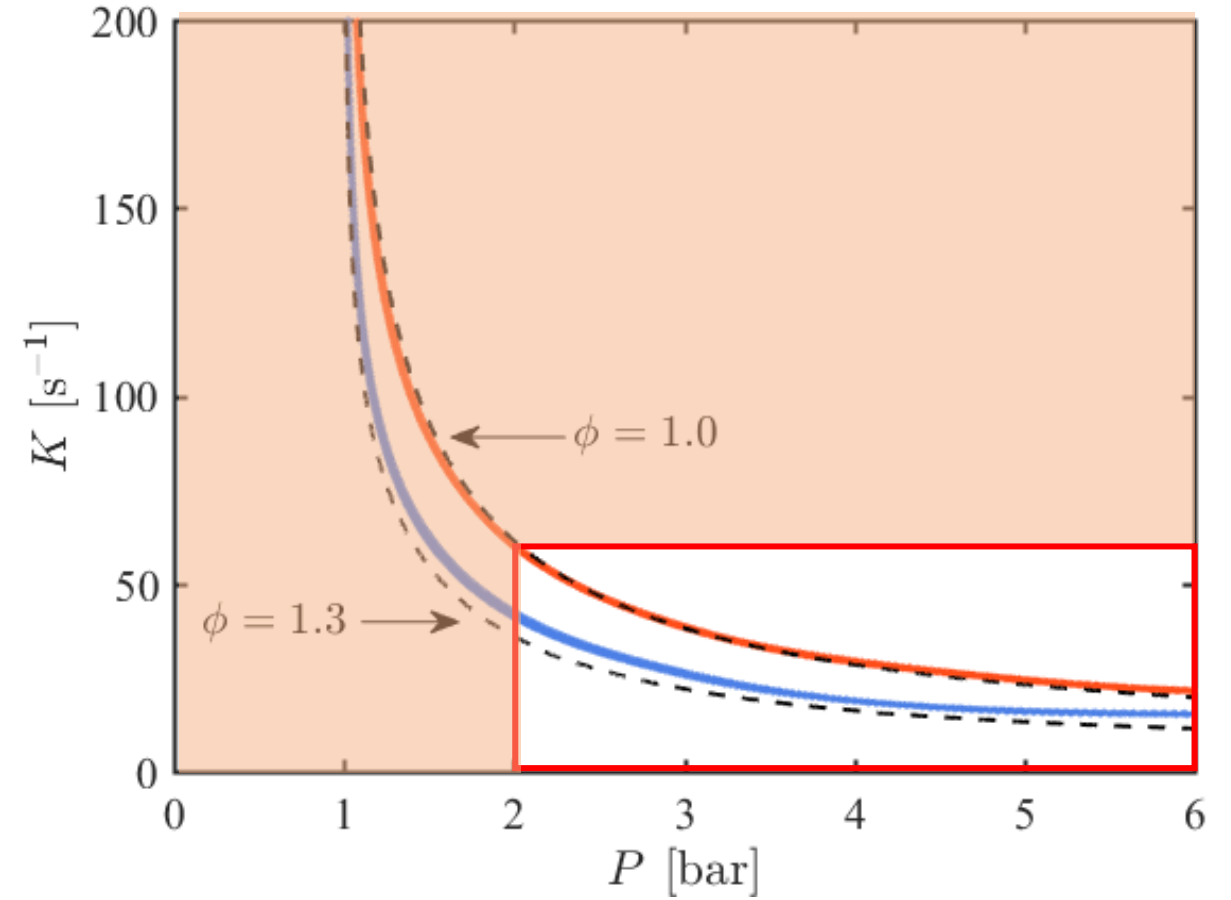
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# Stretch effect

Fuel	$T_0$ (K)	$P_0$ (bar)	$\phi$ (-)	$S_u^0$ (m/s)	$L_u$ (mm)
$CH_4$	300	1	1	0.36	-0.13
$CH_4$	300	1	1.3	0.22	0.3



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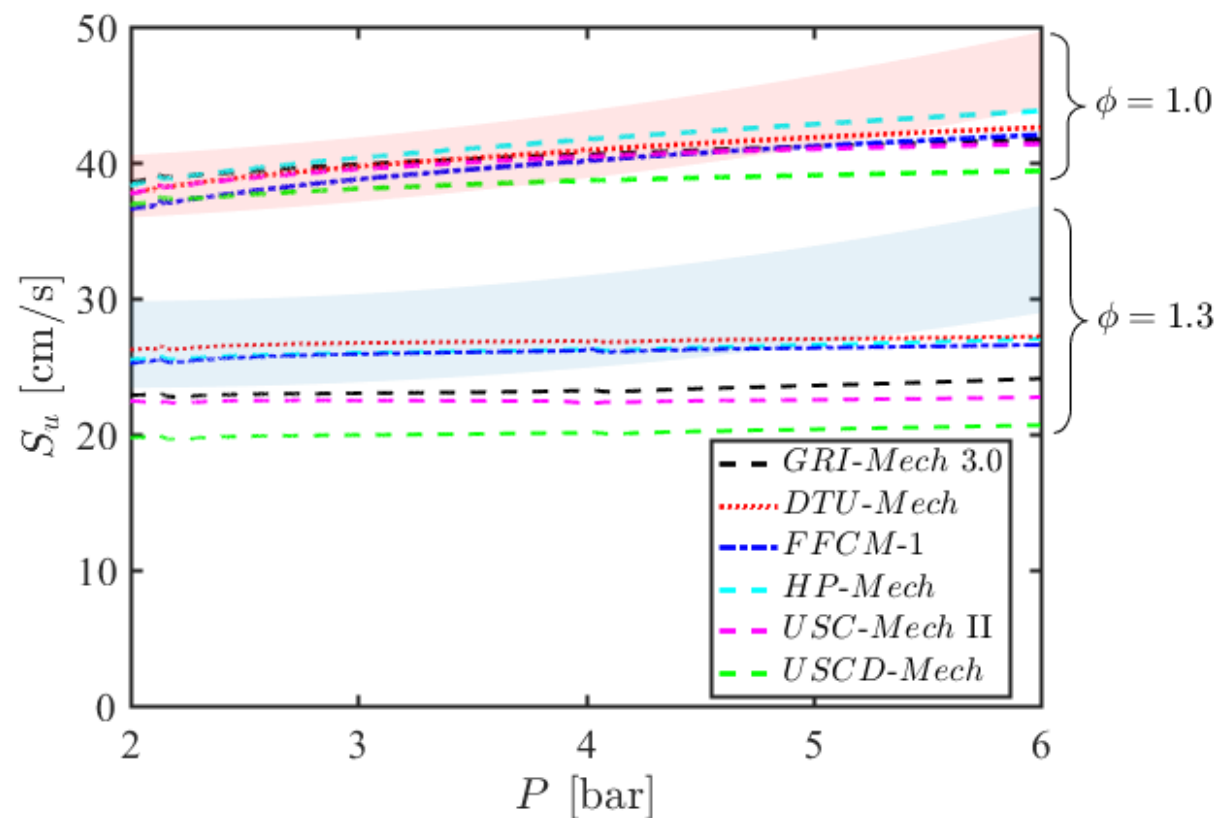
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# Flame speed evaluation

$$S_u = \frac{dR_f}{dt} - \frac{(R_c^3 - R_f^3)}{3\gamma_u R_f^2 P} \frac{dP}{dt}$$

Mechanism	Species	Reactions
GRI Mech 3.0 [1]	53	325
FFCM-1 [2]	38	291
USC Mech II [3]	111	784
UCSD Mech [4]	58	270
DTU Mech [5]	68	631
HP Mech [6]	92	625



[1] G. P. Smith, et al. , [http://www.me.berkeley.edu/gri\\_mech/](http://www.me.berkeley.edu/gri_mech/)

[2] G. P. Smith, et al. , <http://nanoenergy.stanford.edu/ffcm>, (2016)

[3] H. Wang, et al. , [http://ignis.usc.edu/USC\\_Mech\\_II.htm](http://ignis.usc.edu/USC_Mech_II.htm), (2007)

[4] Chemical-Kinetic Mechanisms for Combustion Applications, San Diego Mechanism web page, Mechanical and Aerospace Engineering (Combustion Research), University of California at San Diego (<http://combustion.ucsd.edu>)

[5] H. Hashemi, et al. , High-pressure oxidation of methane, *Combustion and Flame*, 172:349-64 (2016)

[6] <http://engine.princeton.edu/mechanism/HP-Mech.html>,

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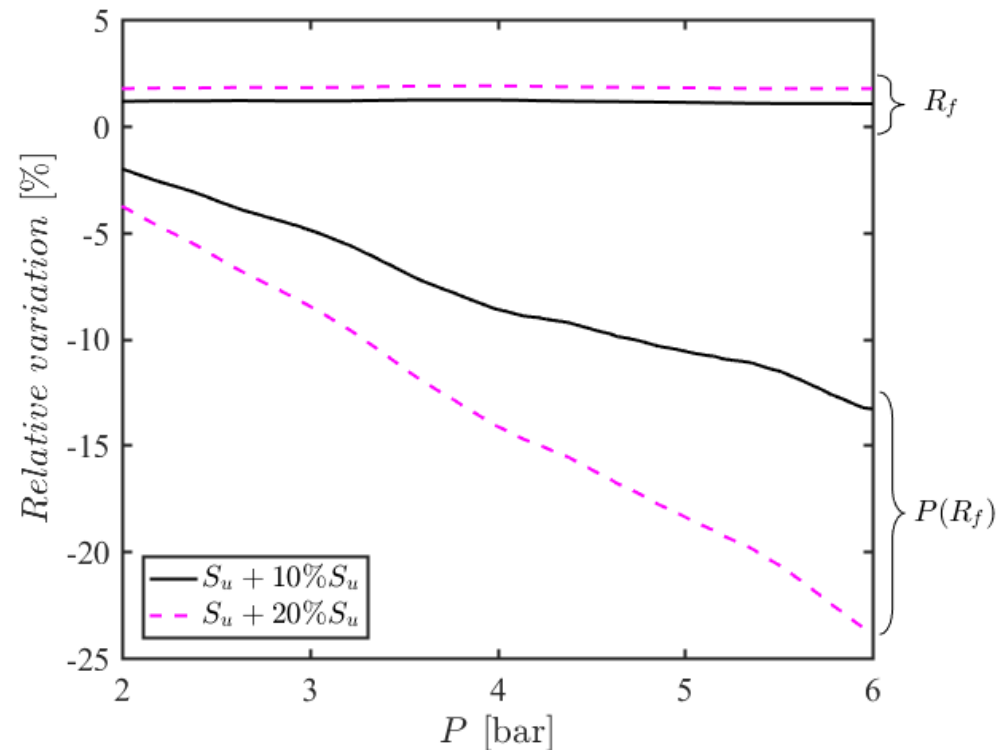
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# Alternative method

$$S_u = \frac{dR_f}{dt} - \frac{(R_c^3 - R_f^3)}{3\gamma_u R_f^2 P} \frac{dP}{dt} \quad \longrightarrow \quad S_{u,n} = \frac{R_{f,n+1} - R_{f,n}}{t_{n+1} - t_n} - \frac{R_c^3 - R_{f,n}^3}{3\gamma_{u,n} P_n R_{f,n}^2} \frac{P_{n+1} - P_n}{t_{n+1} - t_n}$$

$$\begin{cases} R_{f,n+1} = R_{f,n} + S_{u,n} \cdot (t_{n+1} - t_n) + \frac{R_c^3 - R_{f,n}^3}{3\gamma_{u,n} P_n R_{f,n}^2} \cdot (P_{n+1} - P_n) \\ P_{n+1} = P_n + \left( (R_{f,n+1} - R_{f,n}) - S_{u,n} \cdot (t_{n+1} - t_n) \right) \cdot \frac{3\gamma_{u,n} P_n R_{f,n}^2}{R_c^3 - R_{f,n}^3} \end{cases}$$



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# Pressure evolutions

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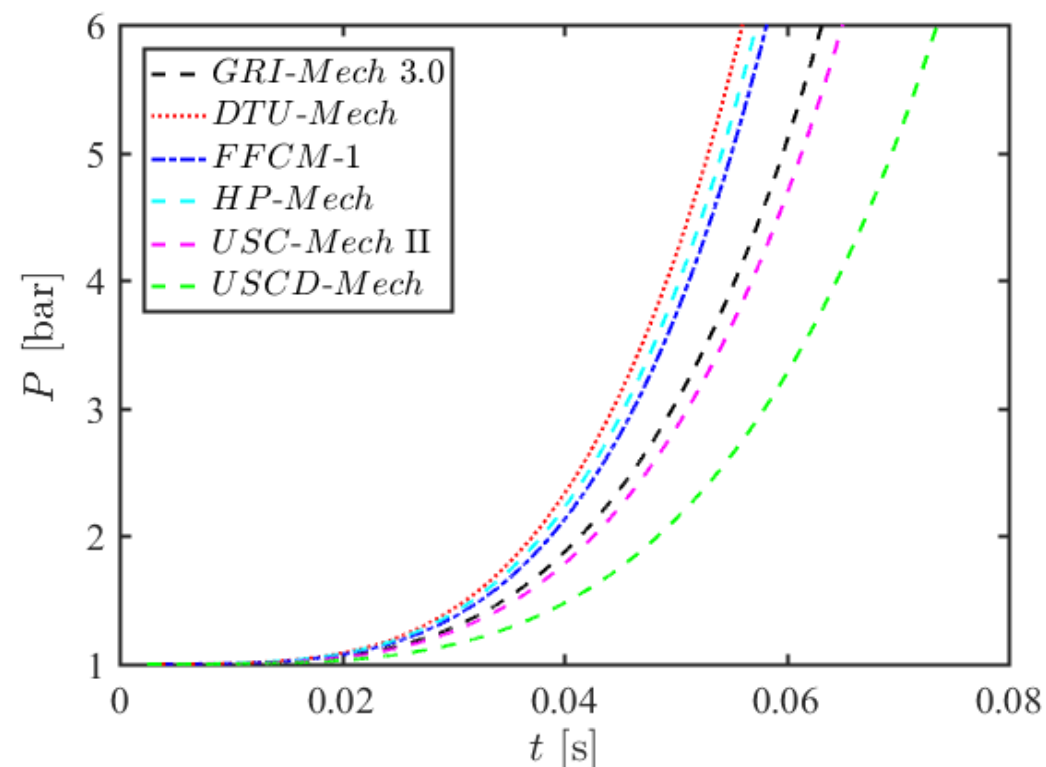
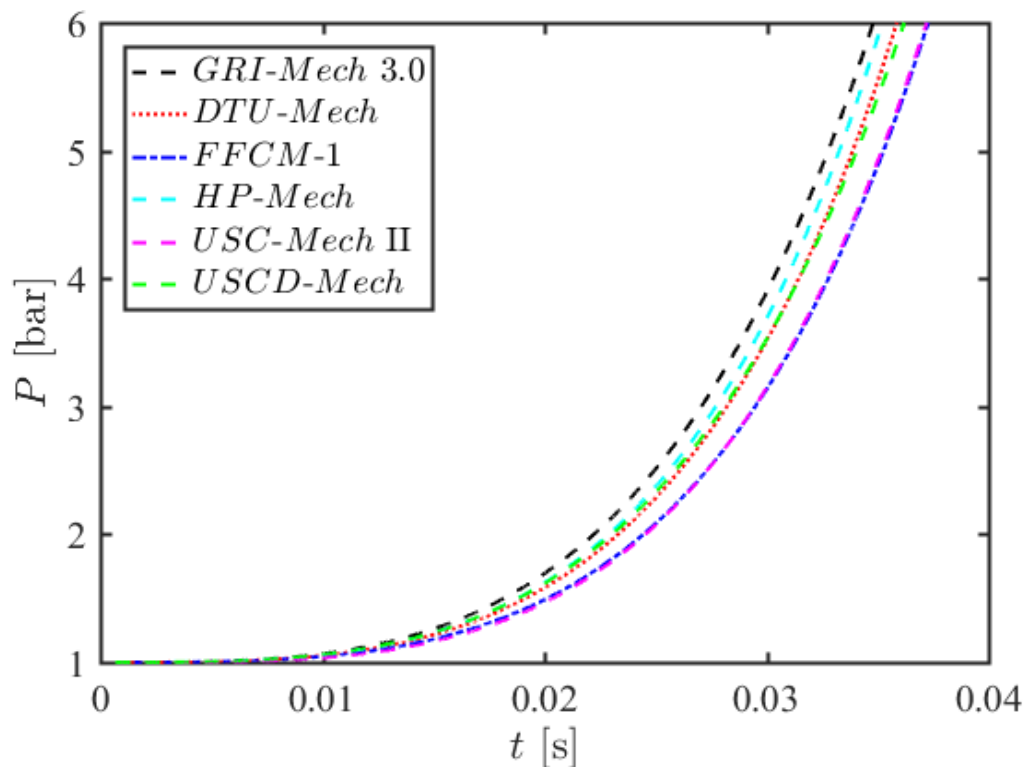
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$$\left\{ \begin{array}{l} P_{n+1} = P_n + \left( (R_{f,n+1} - R_{f,n}) - S_{u,n} \cdot (t_{n+1} - t_n) \right) \cdot \frac{3 \gamma_{u,n} P_n R_{f,n}^2}{R_c^3 - R_{f,n}^3} \\ T_{n+1} = T_n \left( \frac{P_n}{P_{n+1}} \right)^{(1-\gamma_{u,n})/\gamma_{u,n}} \end{array} \right.$$

+ 1D steady simulation

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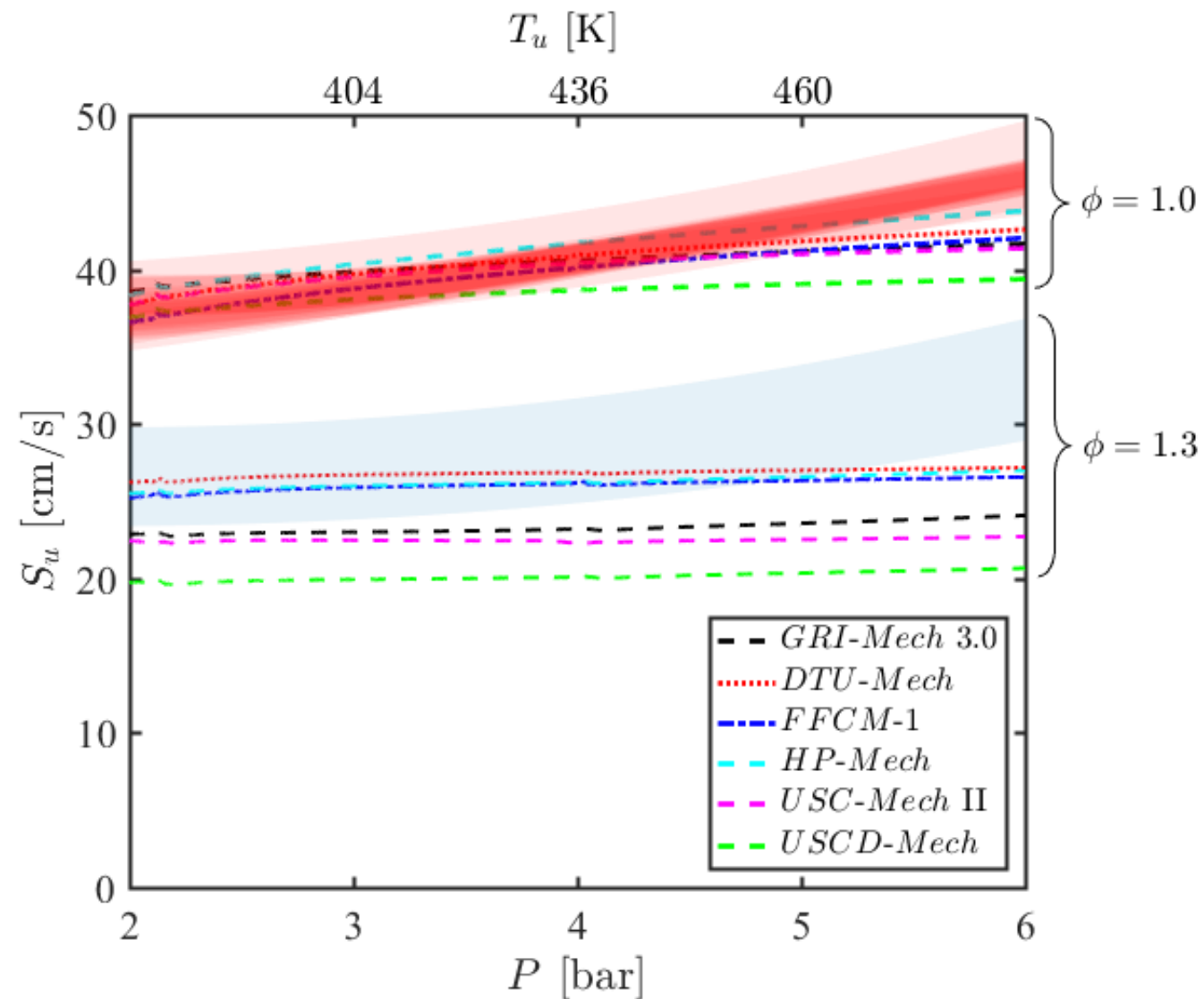
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# Flame speed as a function of pressure

$$P_{n+1} = P_n + \left( (R_{f,n+1} - R_{f,n}) - S_{u,n} \cdot (t_{n+1} - t_n) \right) \cdot \frac{3 \gamma_{u,n} P_n R_{f,n}^2}{R_c^3 - R_{f,n}^3} + S_u = f(P, T)$$



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# Conclusions

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- full **OPTI**cal access **P**erfectly sphe**R**ical combust**I**on cha**M**ber
- **Simultaneous** recording of the **pressure** inside the chamber and, fully innovative, of the flame **radius** until the walls
- Accurate flame speed as a function of pressure/temperature evolution
- **Pressure** is the correct target to assess the accuracy of a kinetic mechanism
- A relative error lower than  $\pm 5\%$  over almost the entire pressure range was obtained
- The unmatched accuracy allows to **optimize** kinetic schemes